

Q.4 a. The resistivity of pure copper is $1.56\mu\Omega\text{-cm}$. An alloy of copper containing 1 atomic per cent nickel has a resistivity of $2.81\mu\Omega\text{-cm}$. An alloy of copper containing 3 atomic percent silver has a resistivity of $1.98\mu\Omega\text{-cm}$. What is the resistivity of an alloy containing 2 atomic percent nickel and 2 atomic percent silver?

Answer:

$$\rho_{cu} = 1.56$$

$$\rho_{(cu+Ni)} = 2.81$$

ρ_I , Increase in resistivity for one atomic percent added impurity (Nickel)

$$\rho_{I(nickel)} = 1.25$$

$$= 2.81 - 1.56 = 1.25$$

$$\rho_{I(silver)} = \frac{1.98 - 1.56}{3} = 0.14$$

$$\rho_{alloy} = 1.56 + 2 \times 1.25 + 2 \times 0.14 = 4.34\mu\Omega - cm$$

TEXT BOOKS

1. Materials Science and Engineering – A First Course by V. Raghavan, Fifth Edition, Thirty-Fourth Print, April 2007 Edition, Prentice-Hall Of India Pvt Ltd.

2. Introduction to Electrical Engineering Materials by C.S. Indulkar and S. Thiruvengadam, 4th Edition, Reprint 2006, S. Chand and Company Ltd.