ROLL NO. \_\_\_\_

## Diplete – Et (NEW SCHEME) – Code: DE54

Subject: ENGINEERING MATERIALS

Time: 3 Hours

## **DECEMBER 2011**

Max. Marks: 100

NOTE: There are 9 Questions in all.

- Please write your Roll No. at the space provided on each page immediately after receiving the Question Paper.
- Question 1 is compulsory and carries 20 marks. Answer to Q.1 must be written in the space provided for it in the answer book supplied and nowhere else.
- The answer sheet for the Q.1 will be collected by the invigilator after 45 Minutes of the commencement of the examination.
- Out of the remaining EIGHT Questions answer any FIVE Questions. Each question carries 16 marks.
- Any required data not explicitly given, may be suitably assumed and stated.

## **Q.1** Choose the correct or the best alternative in the following: $(2 \times 10)$

a. The conductivity of copper is less than that of silver by

( <b>A</b> ) 5-10%	<b>(B)</b> 50% - 60%
( <b>C</b> ) 80%-90%	<b>(D)</b> 20% - 30%

b. The dielectric constant of air is practically taken as

(A) unity	( <b>B</b> ) more than unity
(C) Zero	<b>(D)</b> less than unity

c. Ceramics are good

(A) insulators	(B) conductors
(C) super conductors	(D) Semi conductor

d. Hysteresis loss least depends on

(A) frequency	( <b>B</b> ) Magnetic Field intensity
(C) Volume of material	( <b>D</b> ) Grain orientation of material

e. What is the type of bonding in silicon?

(A) ionic	( <b>B</b> ) covalent
(C) Metallic	<b>(D)</b> Metallic + ionic

f. Copper is completely miscible with

(A) Nickel	( <b>B</b> ) Gold
(C) Hydrogen	(D) Lead

g. No. of terminals in a FET are

(A) one	<b>(B)</b> two
(C) three	<b>(D)</b> four

h. Copper - constanton thermocouple is used for measuring temperature upto

( <b>A</b> ) $1400 {}^{0}$ C	<b>(B)</b> $1100 {}^{0}\text{C}$
$(\mathbf{C}) 800 \ ^{0}\mathbf{C}$	<b>(D)</b> 400 <sup>0</sup> C

i. In a transistor which of the following region is very lightly doped and is very thin?

	<ul><li>(A) Emitter</li><li>(C) collector</li></ul>	<ul><li>(B) Base</li><li>(D) None of the above</li></ul>
j.	Carbon – resistor contains	
	<ul><li>(A) Solid Carbon granules</li><li>(C) Finally divided carbon black</li></ul>	<ul><li>(B) Pulverized coal</li><li>(D) carbon crystals</li></ul>

## Answer any FIVE Questions out of EIGHT Questions. Each question carries 16 marks.

Q.2	a.	Explain the electron gas model of a metal.	(8)
	b.	Explain the effect of temperature on electrical conductivity of metals.	(8)
Q.3	a.	What is permanent dipole moment? Explain in brief.	(8)
	b.	Derive Clausius-Mossotti relation for solid dielectrics due to internal field	. (8)
Q.4	a.	What are the important requirements of good insulating materials? Give examples & their applications.	some (8)
	b.	Discuss various applications of Dielectric materials.	(8)
Q.5	a.	Give the applications and properties of silicon iron alloy and nickel iron al	loy. ( <b>8</b> )
	b.	Give the properties and application of permanent magnetic materials.	(8)
Q.6	a.	Describe the Hall Effect and explain its relation to the magnetic field conductor.	on a (8)
	b.	Write short notes on(i) Einstein relation (between diffusion constant and mobility)(ii) Doping in semiconductors	4+4)

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Q.7	a.	Describe atomic structure of silicon and Germanium.	(8)
	b.	Explain working of SCR based on its two transistor model.	(8)
Q.8	a.	What is voltage-sensitive resistor? Describe in brief different types of vol sensitive resistors.	tage ( <b>8</b> )
	b.	Give applications of powered iron core and ferrite core. (4	+4)
Q.9	a.	What are the various methods by which junctions are fabricated from j single crystal semiconductor?	pure ( <b>8</b> )
	b.	Discuss Epitaxial diffused junction diode and its applications.	(8)